

# NEW STANDARD ACADEMY

Test Type : Unit Test # 01

04-08-2025

Do not open this Test Booklet until you are asked to do so.

## JEE(MAIN): 12<sup>th</sup> Undergoing/Pass Students

Read carefully the Instructions on the Back Cover of this Test Booklet.

### Important Instructions :

1. Immediately fill in the form number on this page of the Test Booklet with Blue/Black Ball Point Pen. Use of pencil is strictly prohibited.
2. The candidates should not write their Form Number anywhere else (except in the specified space) on the Test Booklet/Answer Sheet.
3. The Test Booklet consists of 90 questions.
4. There are three parts in the question paper 1,2,3 consisting of Physics, Chemistry and Mathematics having 30 questions in each subject and each subject having Two sections. (i) Section-I contains 20 multiple choice questions with only one correct option. Marking scheme : +4 for correct answer, 0 if not attempted and -1 in all other cases. (ii) Section-II contains 10 Numerical Value Type questions. Attempt any 5 questions. First 5 attempted questions will be considered for marking. Marking scheme : +4 for correct answer, 0 if not attempted and -1 in all other cases.
5. Use Blue/Black Ball Point Pen only for writing particulars/markings responses on Side -1 and Side-2 of the Answer Sheet. Use of pencil is strictly prohibited.
6. No candidate is allowed to carry any textual material, printed or written, bits of papers, mobile phone any electronic device etc, except the Identity Card inside the examination hall/room.
7. Rough work is to be done on the space provided for this purpose in the Test Booklet only.
8. On completion of the test, the candidate must hand over the Answer Sheet to the invigilator on duty in the Room/ Hall. However, the candidate are allowed to take away this Test Booklet with them.

Name of the Candidate(In Capitals) \_\_\_\_\_

Date of Examination \_\_\_\_\_

Candidate's Signature: \_\_\_\_\_

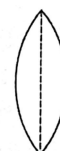
Invigilator's Signature: \_\_\_\_\_

PHYSICS  
SECTION-1

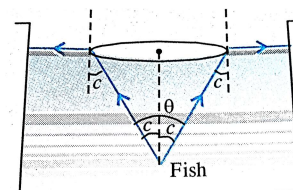
- An object is placed at equal distance (symmetrically) from two mirrors. If the angle of inclination of mirrors is  $40^\circ$ , How many images will be formed?  
(a) 10 (b) 9  
(c) 8 (d) 10.10
- Two mirrors inclined to each other produce five images for an object placed anywhere in between them. The angle between the mirrors is now reduced by  $30^\circ$ . How many images will be observed?  
(a) 11 (b) 12  
(c) 1.1 (d) 1.2
- A plane mirror is rotated at an angular speed of  $3 \text{ rad s}^{-1}$ . Find the angular speed of light being reflected by it.  
(a)  $0.6 \text{ rad s}^{-1}$  (b)  $6 \text{ rad s}^{-1}$   
(c)  $6.6 \text{ rad s}^{-1}$  (d) none of these
- A concave mirror of focal length 20 cm is placed 60 cm from a wall. How far from the wall should an object be placed to form a real image on the wall?  
(a) 20 cm (b) 40 cm  
(c) 30 cm (d) 60 cm
- A square of side 3.0 cm is placed 25 cm away from a concave mirror of focal length 10 cm. What is the area enclosed by the image of wire? (the centre of the wire is on the axis of the mirror, with its two side normal to the axis)  
(a)  $4 \text{ cm}^2$  (b)  $8 \text{ cm}^2$   
(c)  $16 \text{ cm}^2$  (d)  $0.4 \text{ cm}^2$
- If an object approaches towards a plane mirror with velocity  $V$ , then image approaches the object with velocity  
(a)  $V$  (b)  $1.5 V$   
(c)  $2V$  (d)  $3V$
- A ray of light is incident at an angle of  $30^\circ$  on one face of rectangular slab of thickness 8 cm. Find the lateral shift produced if the refractive index of glass is 1.5.  
(a) 1.55 cm (b) 2.55 cm  
(c) 25.5 cm (d) 15.5 cm
- A ray of light passes through a plane boundary separating two media having refractive indices  $\mu_1 = \frac{3}{2}$  and  $\mu_2 = \frac{4}{3}$ . (i) If the ray travels from medium (1) to medium (2) at an angle of incidence  $60^\circ$ , what is the angle of refraction?  
(a)  $50.3^\circ$  (b)  $60.3^\circ$   
(c)  $77^\circ$  (d)  $55.3^\circ$
- We combined a convex lens of focal length  $f_1$  and concave lens of focal lengths  $f_2$  and their

combined focal length was  $F$ . The combination of these lenses will behave like a concave lens if

- $f_1 > f_2$
  - $f_1 < f_2$
  - $f_1 = f_2$
  - $f_1 \leq f_2$
- IN a plano-convex lens the radius of curvature of the convex lens is 10 cm. If the plane side is polished then the focal length will be (Refractive index = 1.5)  
(a) 10.5 cm (b) 10 cm  
(c) 5.5 cm (d) 5 cm
  - A ray of light of frequency  $5 \times 10^{14} \text{ Hz}$  is passed through a liquid. The wavelength of light measured inside the liquid is found to be  $450 \times 10^{-9} \text{ m}$ . Calculate the refractive index of the liquid.  
(a) 1.33 (b) 13.3  
(c) 0.332 (d) 2.33
  - An ink dot on the sheet of a paper is viewed from above a distance of 50 cm. By what distance would the ink dot appear to be raised if it is viewed through a glass slab of thickness 15 cm held parallel to the paper? Given refractive index of glass = 1.5  
(a) 4 cm (b) 5 cm  
(c) 6 cm (d) 7 cm
  - A convex lens has a focal length  $f$ . It is cut into two parts along the dotted line as shown in figure. The focal length of each part will be



- $\frac{f}{2}$
  - $f$
  - $\frac{3}{2}f$
  - $2f$
- The refractive index of water is  $\frac{4}{3}$ . Determine the angle of the cone within which the entire outside view will be confined for the fish under water



- $67.2^\circ$
  - $77.2^\circ$
  - $87.2^\circ$
  - $97.2^\circ$
- A point source of light is situated at the bottom of a tank filled with water up to the level of 80 cm. Find the area of the surface of water through which light from the source emerges out. Take refractive index = 1.33.  
(a) 90.9 cm (b) 60.9 cm

- (c) 30.9 cm                      (d) 60.9 cm
16. If an observer is walking away from the plane mirror with 6 m/sec. Then the velocity of the image with respect to observer will be  
 (a) 6 m/sec                      (b) -6 m/sec  
 (c) 12 m/sec                      (d) 3 m/sec
17. A ray of light is incident at  $50^\circ$  on the middle of one of the two mirrors arranged at an angle of  $60^\circ$  between them. The ray then touches the second mirror, get reflected back to the first mirror, making an angle of incidence of  
 (a)  $50^\circ$                       (b)  $60^\circ$   
 (c)  $70^\circ$                       (d)  $80^\circ$
18. All of the following statements are correct except  
 (a) The magnification produced by a convex mirror is always less than one  
 (b) A virtual, erect, same-sized image can be obtained using a plane mirror  
 (c) A virtual, erect, magnified image can be formed using a concave mirror  
 (d) A real, inverted, same-sized image can be formed using a convex mirror
19. An object of length 6 cm is placed on the principle axis of a concave mirror of focal length  $f$  at a distance of  $4f$ . The length of the image will be  
 (a) 2 cm                      (b) 12 cm  
 (c) 4 cm                      (d) 1.2 cm
20. The focal length of a convex lens is 10 cm and its refractive index is 1.5. If the radius of curvature of one surface is 7.5 cm, the radius of curvature of the second surface will be  
 (a) 7.5 cm                      (b) 15.0 cm  
 (c) 75 cm                      (d) 5.0 cm
21. What will be the height of image when an object of 2 mm is placed on the axis of a convex mirror at a distance 20 cm of radius of curvature 40 cm?
22. Image formed by a concave mirror of focal length 6 cm, is 3 times of the object, then the distance of object from mirror is
23. A concave mirror of focal length  $f$  (in air) is immersed in water ( $\mu = 4/3$ ). The focal length of the mirror in water will be
24. Radius of curvature of concave mirror is 40 cm and the size of image is twice as that of object, then the object distance is
25. A point object is placed at a distance of 10 cm and its real image is formed at a distance of 20 cm from a concave mirror. If the object is moved by 0.1 cm towards the mirror, the image will shift by about
- (c)  $K_2HgI_4 + KOH$                       (d)  $K_2HgI_4 + Hg$
27. Which of the following oxide is white but turns yellow on heating?  
 (a) AgO                      (b) FeO  
 (c) ZnO                      (d) Ag<sub>2</sub>O
28. Which of the following is a strong oxidising agent  
 (a)  $Mn^{3+}$                       (b)  $Zn^{2+}$   
 (c)  $Cr^{3+}$                       (d)  $Sr^{3+}$
29. Electronic configuration of a metal cation  $M^{2+}$  is 2, 8, 14. Number of unpaired electrons in neutral M atom are  
 (a) 4                      (b) 2  
 (c) 5                      (d) 1
30. Main oxidation state shown by lanthanides  
 (a) +3                      (b) +4  
 (c) +2                      (d) +1
31. How many ions are produced by  $[Co(NH_3)_6]Cl_2$  in solution?  
 (a) 6                      (b) 4  
 (c) 3                      (d) 2
32. EDTA is a.....ligand:  
 (a) Monodentate                      (b) Hexadentate  
 (c) Bidentate                      (d) Tridentate
33. The EAN of platinum in potassium hexachloroplatinate (IV) is:  
 (a) 46                      (b) 86  
 (c) 36                      (d) 84
34. The correct IUPAC name of  $Mn_3(CO)_{12}$  is:  
 (a) Dodecacarbonylmanganate (0)  
 (b) Dodecacarbonylmanganic (II)  
 (c) Dodecacarbonyltrimanganese(0)  
 (d) Manganic dodecacarbonyl (0)
35. IUPAC name of complex  $K_3[Al(C_2O_4)_3]$  is:  
 (a) Potassium alumino-oxalate  
 (b) Potassium trioxalatoaluminate (III)  
 (c) Potassium aluminium (III) oxalate  
 (d) Potassium trioxalatoaluminate (VI)
36. Which of the following electronic configuration would be associated with the highest magnetic moment?  
 (a)  $[Ar] 3d^6$                       (b)  $[Ar] 3d^8$   
 (c)  $[Ar] 3d^3$                       (d)  $[Ar] 3d^7$
37. Choose the correct option having all the elements with  $d^{10}$  electronic configuration from the following:  
 (a)  $^{27}Co, ^{28}Ni, ^{26}Fe, ^{24}Cr$   
 (b)  $^{29}Cu, ^{30}Zn, ^{48}Cd, ^{47}Ag$   
 (c)  $^{46}Pd, ^{28}Ni, ^{26}Fe, ^{24}Cr$   
 (d)  $^{28}Ni, ^{24}Cr, ^{26}Fe, ^{29}Cu$
38.  $KMnO_4$  decomposes on heating at 513K to form  $O_2$  along with  
 (a) Mn and  $KO_2$                       (b)  $K_2MnO_4$  and  $MnO_2$   
 (c)  $K_2MnO_4$  and Mn                      (d)  $MnO_2$  and  $K_2O_2$

### CHEMISTRY

26. Nessler's reagent is:

- (a)  $K_2HgI_4$                       (b)  $K_2HgI_2 + KOH$

39. Given below are two statements: one is labelled as Assertion (A) and the other is labelled as Reason (R)  
 Assertion (A): In aqueous solutions  $\text{Cr}^{2+}$  is reducing while  $\text{Mn}^{3+}$  is oxidising in nature.  
 Reason (R): Extra stability to half filled electronic configuration is observed than incompletely filled electronic configuration.  
 In the light of the above statements, choose the most appropriate answer from the options given below:  
 (a) (A) is true but (R) is false  
 (b) Both (A) and (R) are true and (R) is the correct explanation of (A)  
 (c) Both (A) and (R) are true but (R) is not the correct explanation of (A)  
 (d) (A) is false but (R) is true
40. The number of element from the following that do not belong to lanthanoids is  
 Eu, Cm, Er, Tb, Yb and Lu  
 (a) 5 (b) 1  
 (c) 3 (d) 4
41. In which of the following complexes oxidation state of metal is zero?  
 (a)  $[\text{Pt}(\text{NH}_3)_2\text{Cl}_2]$  (b)  $[\text{Cr}(\text{CO})_6]$   
 (c)  $[\text{Cr}(\text{NH}_3)_3\text{Cl}_3]$  (d)  $[\text{Cr}(\text{en})_2\text{Cl}_2]$
42. Among the following which are ambidentate ligand?  
 (i)  $\text{NO}_2^-$  (ii)  $\text{NO}_3^-$   
 (iii)  $\text{EDTA}^{4-}$  (iv)  $\text{C}_2\text{O}_4^{2-}$   
 (v)  $\text{SCN}^-$  (vi)  $\text{H}_2\text{NCH}_2\text{CH}_2\text{NH}_2$   
 (a) (i) and (ii) (b) (iii) and (iv)  
 (c) (i) and (vi) (d) (i) and (v)
43. The IUPAC name of  $[\text{Ni}(\text{NH}_3)_4][\text{NiCl}_4]$  is:  
 (a) tetrachloronickel(II) tetraamminenickel(II)  
 (b) tetraamminenickel(II) tetrachloronickel(II)  
 (c) tetraamminenickel(II) tetrachloronickelate(II)  
 (d) tetrachloronickel(II) tetraamminenickelate(II)
44. Which one of the following is expected to exhibit optical isomerism (en = ethylenediamine)  
 (a)  $\text{cis}[\text{Pt}(\text{NH}_3)_2\text{Cl}_2]$   
 (b)  $\text{trans}[\text{Co}(\text{en})_2\text{Cl}_2]$   
 (c)  $\text{trans}[\text{Pt}(\text{NH}_3)_2\text{Cl}_2]$   
 (d)  $\text{cis}[\text{Co}(\text{en})_2\text{Cl}_2]$
45. Isomerisms exhibited by  $[\text{Cr}(\text{NH}_3)_2(\text{H}_2\text{O})_2\text{Cl}_2]^+$  are  
 (a) Coordinate, linkage  
 (b) Geometrical optical  
 (c) Coordinate geometrical  
 (d) Ionization linkage
46. Among  $\text{CrO}$ ,  $\text{Cr}_2\text{O}_3$  and  $\text{CrO}_3$  the sum of spin – only magnetic moment values of basic and amphoteric oxides is \_\_\_\_\_  $10^{-2}$  BM (nearest integer).
47. Number of colourless lanthanoid ions among the following is \_\_\_\_\_.  
 $\text{Eu}^{+3}$ ,  $\text{Lu}^{+3}$ ,  $\text{Nd}^{3+}$ ,  $\text{La}^{3+}$ ,  $\text{Sm}^{3+}$
48. What is the sum of oxidation number and coordination number of Cr in  $\text{CrCl}_3.6\text{NH}_3$ , is
49. Ethylenediaminetetraacetate ion is a polydentate ligand and negatively charged. The magnitude of negative charge is.
50. How many isomers are possible for the complex ion,  $[\text{Cr}(\text{NH}_3)(\text{OH})_2\text{Cl}_3]^{2-}$ ?
- MATHS**
51. If  $\int \frac{dx}{1+\sin x} = \tan\left(\frac{x}{2} + a\right) + b$ , find the values of a and b.  
 (a)  $\tan\left(\frac{x}{2} + \frac{\pi}{4}\right) + C$  (b)  $\cot\left(\frac{x}{2} + \frac{\pi}{4}\right) + C$   
 (c)  $\tan\left(\frac{x}{2} - \frac{\pi}{4}\right) + C$  (d)  $\tan\left(\frac{x}{2} + \frac{\pi}{4}\right) - C$
52.  $\int \frac{\cos 2x - \cos 2\alpha}{\cos x - \cos \alpha} dx$  is equal to  
 (a)  $2(\sin x + x \cos \alpha) + C$   
 (b)  $2(\sin x - x \cos \alpha) + C$   
 (c)  $2(\sin x + 2x \cos \alpha) + C$   
 (d)  $2(\sin x - 2x \cos \alpha) + C$
53.  $\int \cot x \log(\sin x) dx$  is equal to  
 (a)  $\frac{1}{2} (\log(\sin x))^2 + C$   
 (b)  $\log(\sin x) + C$   
 (c)  $\frac{1}{2} (\log(\cos x))^2 + C$   
 (d) none of these
54.  $\int e^x (1 - \cot x + \cot^2 x) dx$  is equal to  
 (a)  $e^x \operatorname{cosec} x + C$  (b)  $-e^x \operatorname{cosec} x + C$   
 (c)  $e^x \cot x + C$  (d)  $-e^x \cot x + C$
55.  $\int (\sin^4 x - \cos^4 x) dx$  is equal to  
 (a)  $\frac{1}{2} \cos 2x + C$  (b)  $-\frac{1}{2} \cos 2x + C$   
 (c)  $\frac{1}{2} \sin 2x + C$  (d)  $-\frac{1}{2} \sin 2x + C$
56.  $\int e^{3\log x} (x^4 + 1)^{-1} dx$  is equal to  
 (a)  $\frac{1}{4} \log(x^4 + 1) + c$  (b)  $-\frac{1}{4} \log(x^4 + 1) + c$   
 (c)  $\log(x^4 + 1) + c$  (d) none of these
57.  $\int \frac{1+x+\sqrt{x+x^2}}{\sqrt{x}+\sqrt{1+x}} dx$  is equal to  
 (a)  $\frac{1}{2} \sqrt{1+x} + C$  (b)  $\frac{2}{3} (1+x)^{3/2} + C$   
 (c)  $2(1+x)^{3/2} + C$  (d)  $\sqrt{1+x} + C$
58.  $\int \frac{f'(x)}{f(x) \log(f(x))} dx$  is equal to  
 (a)  $f(x) \log(f(x)) + C$   
 (b)  $\log(\log(f(x))) + C$   
 (c)  $\frac{f(x)}{\log(f(x))} + c$

- (d)  $\frac{1}{\log(\log(f(x)))} + c$
59.  $\int e^{x \log a} e^x dx$  is equal to  
 (a)  $\frac{a^x}{\log ae} + C$  (b)  $\frac{e^x}{1+\log}$   
 (c)  $(ae)^x + C$  (d)  $\frac{(ae)^x}{\log ae} + C$
60.  $\int x^{51}(\tan^{-1}x + \cot^{-1}x)dx =$   
 (a)  $\frac{x^{52}}{52}(\tan^{-1}x + \cot^{-1}x) + c$   
 (b)  $\frac{x^{52}}{52}(\tan^{-1}x - \cot^{-1}x) + c$   
 (c)  $\frac{\pi x^{52}}{104} + \frac{\pi}{2} + c$   
 (d)  $\frac{x^{52}}{52} + \frac{\pi}{2} + c$
61.  $\int \left(x + \frac{1}{x}\right)^{3/2} \left(\frac{x^2-1}{x^2}\right) dx$  is equal to  
 (a)  $\frac{1}{3} \left(x + \frac{1}{x}\right)^3 + c$   
 (b)  $\frac{2}{5} \left(x + \frac{1}{x}\right)^{5/2} + c$   
 (c)  $\frac{2}{3} \left(x + \frac{1}{x}\right)^{3/2} + c$   
 (d)  $\frac{4}{3} \left(x + \frac{1}{x}\right)^{3/4} + c$
62.  $\int \frac{x^2 \tan^{-1} x^3}{1+x^6} dx$  is equal to  
 (a)  $\tan^{-1}(x^3) + c$  (b)  $\frac{1}{6} \tan^{-1}(x^3)^2 + c$   
 (c)  $\frac{1}{2} \tan^{-1}(x^3)^2 + c$  (d)  $\frac{1}{2} \tan^{-1}(x^2)^3 + c$
63.  $\int \frac{dx}{\sin \sqrt{3} \cos} =$   
 (a)  $\log \tan \left(\frac{x}{2} + \frac{\pi}{2}\right) + c$   
 (b)  $\frac{1}{2} \log \tan \left(\frac{x}{2} + \frac{\pi}{6}\right) + c$   
 (c)  $\log \cot \left(\frac{x}{2} + \frac{\pi}{6}\right) + c$   
 (d)  $\frac{1}{2} \log \cot \left(\frac{x}{2} + \frac{\pi}{2}\right) + c$
64.  $\int \frac{x-2}{x(2 \log x - x)} dx =$   
 (a)  $\log (2 \log x - x) + c$  (b)  $\log \left(\frac{1}{2 \log x - x}\right) + c$   
 (c)  $\log(x-2 \log x) + c$  (d)  $\log \left(\frac{1}{x-2 \log x}\right) + c$
65. If  $f(x) = \int \frac{x^2 dx}{(1+x^2)(1+\sqrt{1+x^2})}$  and  $f(0) = 0$  then the value of  $f(1)$  will be  
 (a)  $\log (1+\sqrt{2})$  (b)  $\log (1+\sqrt{2}) - \frac{\pi}{4}$   
 (c)  $\log (1+\sqrt{2}) + \frac{\pi}{2}$  (d) None of these
66. The value of  $\int e^x(x^5 + 5x^4 + 1)dx$  is equal to  
 (a)  $e^x x^5 + e^x + c$  (b)  $e^x x^5 + c$   
 (c)  $5x^4 e^x + c$  (d)  $e^{x+1} x^5 + c$
67.  $\int \frac{\sin^8 x - \cos^8 x}{1 - 2 \sin^2 x \cos^2 x} dx =$   
 (a)  $(\sin 2x)/2 + c$  (b)  $-(\sin 2x)/2 + c$   
 (c)  $-(\sin x)/2 + c$  (d)  $-\sin^2 x + c$
68.  $\int \sin x \cdot \cos x \cdot \cos 2x \cdot \cos 4x \cdot \cos 8x dx$  is equal to  
 (a)  $-\frac{1}{96} \cos 16x + c$  (b)  $-\frac{1}{256} \cos 16x + c$
- (c)  $-\frac{1}{16} \cos 16x + c$  (d) None of these
69. If  $\int \frac{\cos x - \sin x + 1 - x}{e^x + \sin x + x} dx = \log_e(f(x)) + g(x) + c$  where C is the constant of integration and f(x) is positive, Then f(x) + g(x) has the value equal to  
 (a)  $e^x + \sin x + 2x$  (b)  $e^x + \sin x$   
 (c)  $e^x - \sin x$  (d)  $e^x + \sin x + x$
70.  $\int \frac{\ln \left(\frac{x-1}{x+1}\right)}{x^2-1} dx$  is equal to  
 (a)  $\frac{1}{2} \left(\ln \left(\frac{x-1}{x+1}\right)\right)^2 + C$  (b)  $\frac{1}{2} \left(\ln \left(\frac{x+1}{x-1}\right)\right)^2 + C$   
 (c)  $\frac{1}{4} \left(\ln \left(\frac{x-1}{x+1}\right)\right)^2 + C$  (d)  $\frac{1}{4} \left(\ln \left(\frac{x+1}{x-1}\right)\right)^2 + C$
- Numerical value type**
71. If  $\int \frac{\cos 4x}{\cot x - \tan x} dx = 4 \cos 4x + b$ , then the value of |A| is \_\_\_\_\_
72. Let  $f(x) = \int x^{\sin x} (1 + x \cos x \ln x + \sin x) dx$  and  $f\left(\frac{\pi}{2}\right) = \frac{\pi^2}{4}$ . Then the value of  $f(\pi)$  is \_\_\_\_\_
73. If  $\int \left[\left(\frac{x}{e}\right)^x + \left(\frac{e}{x}\right)^x\right] \ln x dx = A \left(\frac{x}{e}\right)^x + B \left(\frac{e}{x}\right)^x + C$ , then the value of |AB| is \_\_\_\_\_.
74. Let  $f(x) = \int \frac{(x^2+1)dx}{\sqrt[3]{x^3+3x+6}}$  and  $f(-1) = \frac{1}{\sqrt[3]{2}}$ . Then the value of  $f(-2)$  is \_\_\_\_\_
75. If  $\int \frac{2 \cos x - \sin x + \lambda}{\cos x + \sin x - 2} dx = A \ln |\cos x + \sin x - 2| + Bx + C$ , then the value of |AB| is \_\_\_\_\_